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energy  $h\nu$ , where  $\nu$  is the frequency of radiation. The individual quanta of light are called photons. Einstein's photoelectric equation  $h\nu = h\nu_0 + \frac{1}{2}mv^2$  (1.4) explained all aspects of photoelectric effect. In Eq. (1.4),  $h\nu$  is the energy of the incident photon,  $h\nu_0$

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