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kinetic energy. The energy an object has because of its motion. kinetic theory. All matter consists of tiny particles that are in constant motion. molecules/atom. the particles in a gas usually consist of. insignificant. The particles in a gas are considered to be small, hard spheres with an _____ volume. empty space ...

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constant motion. molecules/atom. the particles in a gas usually consist of. empty space. what is between particles in a gas? rapid, constant, and random. the motion of the particles in a ...

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Chapter 6. alkali metals. alkaline earth
metals. halogen. cation. any metal in
Group 1A of the periodic table. any metal
in Group 2A of the periodic table. a
nonmetal in Group 7A of the periodic
table. any atom or group of atoms with a

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positive charge . alkali metals. any metal
in Group 1A of the periodic table ...

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of kinetic theory as applied to gases-The
particles in a gas are considered to be
small, hard spheres with an insignificant
volume-The motion of the particles in a gas
is rapid, constant, and random ...

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Hall Chemistry Chapter 12 Vocabulary.
Allotrope. Amorphous Solid. Atmospheric
Pressure. Barometer. Two or more
different molecular forms of the same

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element in t... A solid without an ordered internal structure. The result from the collisions of atoms and molecules in air w... A device that is used to measure atmospheric pressure. Allotrope. Two or more ...

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kinetic energy. kinetic theory.

molecules/atom. insignificant. The energy an object has because of its motion. All matter consists of tiny particles that are in constant mot... the particles in a gas usually consist of. The particles in a gas are

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considered to be small, hard sphere...
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Hall Chemistry Chapter 12 Vocabulary.
Allotrope. Amorphous Solid. Atmospheric Pressure. Barometer. Two or more different molecular forms of the same element in t... A solid without an ordered internal structure. The result from the collisions of atoms and molecules in air w... A device that is used to measure

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atmospheric pressure. Allotrope. Two or more ...

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STUDY. PLAY. Kinetic Energy. The
energy an object has because of its motion.

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Kinetic Theory . The theory that states that all matter consists of tiny particles that are in constant motion. When applied to gases makes the following fundamental assumptions: The particles in a gas are small hard spheres with insignificant volume, The motion of gas particles is ...

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edition Theodore L. Brown; H. Eugene
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Chapter 7 Ionic and Metallic Bonding.

155. Section Review. Objectives. $\hat{\epsilon}\phi$
Determine the number of valence ... D^1

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acquire the stable electron structure of a
noble gas by losing electrons. ... Answer
the following in the ...

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