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(Video 1) **A Tour of the**

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color and label the 10

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element groups shown on the web site. Be sure to create a 'key' so you know which color represents which group!!

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1. What does the atomic
number of an element
represent? _____ .36 2. If

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an atom is neutral, then the number of electrons is equal to the number of _____ .

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On your periodic table, draw

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the black stair-step line that distinguishes metals from nonmetals. b.

Metalloids have properties of both c. Define semiconductor: d. Name two metalloids that are semiconductors. t, and e. This

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property makes metalloids
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information asked for. Part I: "Get Organized Periodically" Go to <http://www.chem4kids.com/files/elementtable.html> 1. Why are the elements placed in specific places on the Periodic Table? 2. Periods are _____

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that run from left to right.
3.

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On your own periodic table:
Visit this site: http://go.hrw.com/resources/go_sc/periodic

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dic/SSHK1PER.PDF. In an empty area above the table make a diagram (box) that shows: Where the element symbol is. Where atomic number is located . Where atomic mass is located. Add any new elements that are

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not present on your table.

Number the groups (vertical columns) 1-18

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periodic table was published
by French Academy of Science

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. created fully functioning periodic system on a cylinder . table was arranged with the elements given in numbers . compiled a periodic table of 56 elements based on properties . published his periodic

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table and law in 1869. table was recored according to atomic numbers

*THE PERIODIC TABLE
WEBQUEST!! ON THE SHOULDERS
OF GIANTS*

The Periodic Table (PT) is

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Organized into four main regions. These regions are called metals, nonmetals, metalloids, and Noble Gases. These regions are broken down even further into vertical columns called Groups or Families. Not only

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is the PT vertically
organized, it's got it all
together horizontally, too!

*Periodic Poster/Element
Demonstration Project*
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information asked for. Part
I: "Major Players" Go to....

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development of the Periodic
Table of Elements.

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Elcho Table*

Elements as Building Blocks.
The periodic table is
organized like a big grid.

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Each element is placed in a specific location because of its atomic structure. As with any grid, the periodic table has rows (left to right) and columns (up and down). Each row and column has specific

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