#### Modern Chemistry Chapter 4 Test

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The Periodic Table: Crash Course Chemistry #4

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1s 22s 2p 63s23p 3d104s1 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 9. 1 1012 m What is the wavelength of light that ...

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Modern Chemistry 105 Chapter Test Name Class Date Chapter Test A, continued Use this figure to answer questions 7 and 8. \_\_\_\_\_ 7. A solution containing 35 g of Li 2SO 4 dissolved in 100 g of water is heated from 10 ° C to 90 ° C. According to information in the figure, this temperature change would result in a. an additional 5 g of Li 2SO 4 in ...

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Modern Chemistry 27 Chapter Test Name Class Date Chapter Test A, continued \_\_\_\_\_ 4. The electron configuration below violates a. the Pauli exclusion principle. b. the Aufbau principle. c. Hund 's rule. d. Both (a) and (c) \_\_\_\_\_ 5. A photon is emitted from a gaseous atom when an electron moves to its ground state from a(n) a. inner shell. b ...

Assessment Chapter Test A - Ed W. Clark High School Modern Chemistry 33 Chapter Test Name Class Date Chapter Test B, continued 15. The energy state of an atom is called its ground state. 16. The number of waves that pass a point in one second is called. 17.

When an electron drops from a higher-energy state to a lower-energy state, a(n) spectrum is produced. 18.

Assessment Chapter Test B - Ed W. Clark High School File Name: Modern Chemistry Chapter 4 Test.pdf Size: 5133 KB Type: PDF, ePub, eBook Category: Book Uploaded: 2020 Nov 19, 02:18 Rating: 4.6/5 from 758 votes.

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Answers 5 The Periodic Law TEST B 1. a 2. c 3. d 4. d
5. a 6. a 7. c 8. a 9. lanthanides 10. 2 11. fourth 12.
transition elements 13, 32, 14. valence electrons 15.

electron affinity 16. electronegativity 17. ionization energy 18. 3 s2 3p4 19. atomic radius 20 ...

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