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3-dimensional
region around the
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indicates the
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Electrons in Atoms.

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exist. in between
orbits. what are
atomic energy
levels ...

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PROBLEMS Write
the answer on the
line to the left.

Show all your work
in the space
provided. 9. 1 10^{12}
m What is the
wavelength of light
that ...

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105 Chapter Test

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Chapter Test A,

continued Use this

figure to answer

questions 7 and 8.

_____ 7. A solution

containing 35 g of

Li_2SO_4 dissolved

in 100 g of water is

heated from 10°C

to 90°C . According

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to information in
the figure, this
temperature

change would
result in a. an
additional 5 g of Li
2SO 4 in ...

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Chapter Test A,

continued _____ 4.

The electron

configuration below

violates a. the Pauli

exclusion principle.

b. the Aufbau

principle. c. Hund's

rule. d. Both (a)

and (c) _____ 5. A

photon is emitted

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from a gaseous atom when an electron moves to its ground state from a(n) a. inner shell. b ...

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Chapter Test B, continued 15. The energy state of an

atom is called its
ground state. 16.

The number of
waves that pass a
point in one second
is called. 17. When
an electron drops
from a higher-
energy state to a
lower-energy state,
a(n) spectrum is

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produced. 18.

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Answer Key by
Holt, Rinehart,
Winston and a ...

Chapter 26 Nursing
School Test Banks -
Test ... 4. The clinic
nurse has been
assigned to a
43-year-old man
who is overweight
and loves to eat.

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B 1. a 2. c 3. d 4. d

5. a 6. a 7. c 8. a 9.

lanthanides 10. 2

11. fourth 12.

transition elements

13. 32 14. valence

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electrons 15.

electron affinity 16.

electronegativity

17. ionization

energy 18. $3s^2$

$3p^4$ 19. atomic

radius 20 ...

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