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Chemistry - Unit 6 Worksheet 1 We have
observed evidence that when M-NM compounds
are dissolved, the metal particles tend to

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form positively charged ions (cations), while non-metal particles tend to form negatively charged ions (anions).

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Worksheet 5 IONS Representing Ions and Formula Units FORMULA UNIT 1. Ca²⁺ and Br⁻ 2. Fe²⁺ and Cl⁻ 3.

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Modeling Chemistry U6 Ws 3 ©Modeling Instruction-AMTA 1 U6 ws 3 v3.0 Name Date Pd Unit 6 - Worksheet 3 Ionic Compounds Properties Ionic compounds generally have higher mp's and bp's than do molecular compounds. When molten, they conduct electricity. If soluble, the solutions also conduct electricity.

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the formula, give the name of the following simple binary ionic compounds. a. Na_2O b. MgCl_2 c. CaBr_2 d. Al_2S_3 e. CsBr f. AgF g. Na_3N h. K_2O 5. Write the name of these ionic substances next to the formula, using a Roman numeral to specify the charge of the cation. ... Modeling Chemistry U6 Ws 5 V1 0 Agunot

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