

Introductory Chemical Engineering Thermodynamics Solutions Manual

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Equation (6.66b) therefore becomes: $G^R = Z - 1 - \ln(Z q \partial RT - \partial) - Z$ For given T and P, Z is found by solution of Eq. (3.52) for a vapor phase or Eq. (3.56) for a liquid phase with $\sigma = \delta = 0$.

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