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Chem 101 2020-04-08 Sections 5-5 to 5-9 Chem 101 2020-03-30 Sections 4-1 to 4-5 Chem 101 2020-04-02 Sections 4-8 to 4-9 Chem 101 2020-04-20 Sections 7-1 to 7-2 Chem 101 2020-04-14 Sections 6-1 to 6-2 Chem 101 2020-04-28 Sections 7-5 to 7-6 Chem 101 2020-05-04 Sections 7-12 Chem 101 2020-04-07 5-3 to 5-4 Chem 101 2020-04-27 Sections 7-2 to 7-5 Chem 101 2020-04-06 Sections 4-9 to 5-3 Chem 102 2020-04-07 Sections 15-2 to 15-4 Chem 101 2020-04-15 Sections 6-2 to 6-3 Kinetics Lab Perturbation Theory in Quantum Mechanics - Cheat Sheet Statue of Liberty's original torch moved to new museum home How Its Made - 092

Bronze Sculptures AP Chemistry Unit 3
Review: Intermolecular Forces and
Properties Deriving 1st Order
Perturbation Theory (Energy and
Wavefunction Corrections) AP Chemistry
Unit 4 Review: Chemical Reactions AP
Chemistry Unit 5 Part 2 Review: Kinetics
:D

L1.1 General problem. Non-degenerate perturbation theory Conserving Bronze: The Lamp with Erotes from Vani Chem 102 2020-04-15 Section 16-1 Chem 101 2020-05-04 Sections 8-1 to 8-4 Chem 102 2020-03-30 Sections 14-3 to 14-5 Chem 102 2020-05-07 Sections 18-5 to 18-8 Chem 102 2020-05-19 Sections 21-1 to 21-3 Chem 102 2020-04-28 Sections17-2 to 17-5 Chem 102 2020-05-13 Sections 22-1 to 22-2

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#### **General Chemistry**

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=  $c = 2.9979 \times 108 \text{ m} \cdot \text{s} - 1656.3 \times 10 - 9 \text{ m} = 4.568 \times 1014 \text{ s} - 1 \text{ The}$ energy is given by Equation 4.3 E = h =  $(6.626 \times 10 - 34 \text{ J} \cdot \text{s})(4.568 \times 1014 \text{ s} - 1)$ =  $3.027 \times 10 - 19 \text{ J} = 0.3027 \text{ aJ} 4-6$ . We use Equation 4.7 = h mv =  $6.626 \times 10 - 34 \text{ J} \cdot \text{s} (45.9g) \text{ 1kg} 1000 \text{ g} (120 \text{ mi} \cdot \text{h} - 1)$ 

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