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Prentice hall chemistry book. "CORE TEACHING RESOURCES" also known as -CORE Teaching Resources Prentice Hall Chemistry - Spiral-bound - Teacher's Edition (2005) by Antony C. Wilbraham-

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Algebra 1: Common Core (15th Edition) Charles, Randall I. Publisher Prentice Hall ISBN 978-0-13328-114-9

Textbook Answers | GradeSaver

138 Core Teaching Resources SECTION 6.2 CLASSIFYING THE ELEMENTS 1. Use the periodic table to write the electron configuration for silicon. Explain your thinking. 2. Use the periodic table to write the electron configuration for iodine. Explain your thinking. 3. Which group of elements is characterized by an s2p3 configuration? 4.

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This lesson aligns with the Common Core State Standards: STANDARD 5 - Matter and Energy "The teacher understands the nature and properties of energy in its various forms, and the processes by which energy is exchanged and/or transformed. 5A. understands the atomic and nuclear structure of matter and

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