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point compared to oleic acid,
 $\text{CH}_3(\text{CH}_2)_7\text{CH}=\text{CH}(\text{CH}_2)_7\text{COOH}$.

Outline your reason. [2]..... B3. (a) Some synthetic hormones are similar in structure to progesterone and estrogen and may be used to prevent pregnancy. Outline the mode of action of these hormones as oral contraceptives. [3].....

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3. a ionic 1 3. b i red 1 3. b ii emission spectra of both « ^6Li and natural Li» give same colour/produce same «range of»

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wavelengths OR they have same electron transitions/same nuclear charge Accept “the spectra are almost identical”. 1 3. b iii ICP -MS Accept “MS/mass spectrometry”. 1 3. c r m 0.694 n 0.100 mol M 6.94 «

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(vi) $\text{HCOOC}(\text{CH}_3)_3$ is an isomer of X.
For the ^1H NMR spectrum of this isomer,
deduce the total number of peaks
(excluding the TMS peak at 0 ppm) and
the ratio of peak areas. For each peak,
deduce whether it is a singlet, doublet,
triplet, quartet or shows a more complex

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splitting pattern. [3] Number of peaks
(excluding the TMS peak at 0 ppm):

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topics (a total of 95 hours), and HL additionally covers topics 12 to 21 (an additional 60 hours). For both levels, you'll also study one of the four options A-D (15 hours for SL, 25 hours for HL).

Topic #1: Stoichiometric

Relationships—13.5 Hours for SL and HL

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volume made up to the graduated mark with distilled water. A 10.0 cm^3 sample of this new solution was pipetted into a new conical flask and titrated with $0.100 \text{ mol dm}^{-3}$ sodium hydroxide solution using phenolphthalein as the indicator. 12.60

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cm³ of the 0.100 mol dm⁻³ sodium hydroxide solution was required to reach the end point.

SL & HL Paper 3 Section A Experimental work Question 7

The 1:2 ratio of marks (paper 3-to-paper 2) for SL versus the 2:3 ratio for HL is

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important to bear in mind when deciding how much time to give over to each exam in preparation. On a similar note, the emphasis on the Option is different for SL and HL. The Option forms a much smaller fraction of the overall mark if you are studying it at SL.

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