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STUDY. PLAY. single-displacement.
the replacement of hydrogen from
water by sodium is an example of a
_____ reaction. activation energy. in
order for a chemical reaction to take
place, the particles invoved must
collide with a sufficient amount of

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1) calculate moles of each element (if
given mass) 2) divide each mole
quantity by the smallest mole value
from step 1. 3) use whole number
values from step 2 as the subscripts to
the original equation to write the
empirical formula. Percent

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Composition. -the relative amounts of each element in a compound.

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Chemistry - 2nd Semester Practice Test Thermochemistry (Ch.11): Multiple Choice 24) What is the amount of heat required to raise the temperature of 200.0 g of aluminum by 10 °C? cal (specific heat of aluminum = 0.21 J/g °C 24) A) 4200 cal B) 420 cal C) 42 000 cal D) 420 000 cal E) none of the above

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2nd Semester Chemistry-575 Final Exam Review Multiple Choice Identify the choice that best completes the statement or answers the question.

____ 1. How many moles of tungsten atoms are in 4.8×10^{25} atoms of

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tungsten? a. 8.0×10^2 moles c. 1.3×10^{-1} moles b. 8.0×10^1 moles d. 1.3×10^{-2} moles ____ 2.

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Chapter 2 History of chemistry before 1701 Chemistry had a bad reputation because of alchemy, which was a pseudoscience that claimed to convert

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cheap metals into gold. Alchemy led to the discovery of many elements.
1500 – George Bauer discovered metallurgy, extracting pure metal out of a mix
1600s – Robert Boyle studied the behavior of gases and is responsible for the theory of elements
...

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