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Bases By: Loreley Estrada Stephanie
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Preparation Acids And Bases Chapter
19

An acid is a substance that contains hydrogen and ionizes to produce hydrogen ions in aqueous solution; A base is a substance that contains a hydroxide group and dissociates to

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produce a hydroxide ion in aqueous solution.

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Section 19.3 – Strengths of Acids and Bases In general, strong acids completely dissociate in aqueous solution. Weak acids only slightly ionize in aqueous solution. Strong acids include HCl, HNO₃, and H₂SO₄.

Chapter 19 – Acids, Bases, and Salts

Acids are hydrogen containing compounds, ionize to yield hydrogen ions in aqueous solutions. Bases are compounds that ionize to yield hydroxide ions in aqueous solutions. Monoprotic acids. acids that contain 1 ionizable hydrogen such as nitric acid (HNO₃) diprotic acids.

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accept a positive hydrogen ion. Thus,
an acid-base reaction in the reverse
direction can occur. The acid H_3O^+
can react with the base X to form
water and HX and the following
equilibrium is established. 598

Chapter 19 Acids and Bases $\text{HX}(\text{aq}) + \text{H}_2\text{O}(\text{l}) \rightleftharpoons \text{H}_3\text{O}^+(\text{aq}) + \text{X}^-(\text{aq})$
Acid Base
Conjugate acid Conjugate base

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STUDY. PLAY. how did Arrhenius
define an acid and a base? according
to Arrhenius, acids are hydrogen-

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containing compounds that ionise to yield hydrogen ions (H^+) in aqueous solution bases are compounds that ionise to yield hydroxide ions (OH^-) in aqueous solution.

chapter 19: acids, bases, and salts

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by: Gingras, Shelley L. (ASD-E) Created
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Chapter 19 Acids and Bases - Mrs.
Gingras' Chemistry Page
Brønsted-Lowry acids and bases come
in pairs. A conjugate base is the
remainder of the B-L acid, after it
donates its H^+ . A conjugate acid is the
substance formed when the B-L base
gains a H^+ . Thus, a conjugate acid-
base pair is related by the . loss or
gain. of a single hydrogen ion.

Chapter 19 Acids, Bases, and Salts
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What are the properties of acids
and bases? acids taste sour, will
change the color of an acid-base
indicator, and can be strong or weak

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Section 19.1 Acid-Base Theories

OBJECTIVES:

Compare and contrast acids and bases as defined by the theories of: a) Arrhenius, b) Brønsted-Lowry, and c) Lewis.

4.

Chemistry - Chp 19 - Acids, Bases, and
Salt - PowerPoints

Chapter 19: Acids and Bases You ' re probably already aware from what you ' ve learned in other classes that acids and bases are all over the place. From the lemon you put in iced tea to the Drano put down the drain to keep your hair from clogging the drain, acids and bases are an important part

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